organic papers

Acta Crystallographica Section E Structure Reports Online

ISSN 1600-5368

Naotake Nakamura* and Akira Ohishi

Department of Applied Chemistry, College of Science and Engineering, Ritsumeikan University, 1-1-1 Nojihigashi, Kusatsu, Shiga 525-8577, Japan

Correspondence e-mail: nakamura@se.ritsumei.ac.jp

Key indicators

Single-crystal X-ray study T = 296 KMean $\sigma(\text{C-C}) = 0.005 \text{ Å}$ R factor = 0.038 wR factor = 0.116 Data-to-parameter ratio = 16.6

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

1,20-Dibromoicosane

The molecule of the title compound, $C_{20}H_{40}Br_2$, is centrosymmetric and the skeleton has an all-*trans* conformation including both terminal Br atoms. In the crystal structure, the molecules form a layer in which the long axes of the molecules are inclined to the basal plane of the Br atoms. The layers are arranged in a zigzag manner between the neighboring layers, forming a herringbone motif, just like the smectic C_A phase of liquid crystals.

Received 22 June 2004 Accepted 16 July 2004 Online 24 July 2004

Comment

Normal long-chain aliphatic compounds, for example *n*-alkanes, have been studied to elucidate the principles of organic chemical crystallography and basic polymer science, because the molecular skeleton consists of a simple *trans* zigzag straight hydrocarbon chain. The molecular shape of these compounds can be regarded as rod-like, and the molecules in the crystalline state form a layered structure similar to those of the smectic liquid crystalline phase. Moreover, some of these compounds exhibit a high-temperature rotator phase just below their melting points, in which molecules have motional freedom to some degree as well as that in liquid crystals. As a result, normal long-chain aliphatic compounds have also been investigated as models for smectic liquid crystals.



In these investigations, it is important to obtain detailed crystallographic data. Many researchers have analyzed the crystal structures of many different kinds of normal long-chain aliphatic compounds, for example, n-alkanes (e.g. Nyburg & Gerson, 1992), n-primary alcohols (e.g. Michaud et al., 2000), and α, ω -disubstituted *n*-alkanes, such as 1,12-dibromododecane (Kulpe et al., 1981) and 11-bromoundecan-1-ol (Rosen & Hybl, 1972). Recently, we have systematically analysed the crystal structures of the alkane- α, ω -diols containing 10–19 and 21-23 C atoms (Nakamura et al., 2001; Uno et al., 2002), and we have studied the phase-transition phenomena of the series of the alkane- α,ω -diols containing 13–24 C atoms (Ogawa & Nakamura, 1999). In the present paper, we report the crystal structure analysis of the title compound, (I), in order to elucidate the effect of the terminal groups in normal longchain aliphatic compounds.

The molecular structure of (I) is shown in Fig. 1. The molecule is centrosymmetric and all torsion angles are close to $\pm 180^{\circ}$, that is, the molecular skeleton including both terminal

© 2004 International Union of Crystallography

Printed in Great Britain - all rights reserved



Figure 1

The molecular structure of (I), showing the atom numbering scheme. Displacement ellipsoids are drawn at the 50% probability level. [Symmetry code: (i) 2 - x, -y, -z.]



Figure 2 The projection of the crystal structure of (I) along the b axis.



Figure 3 The projection of the crystal structure of (I) along the a axis.

Br atoms has an all-*trans* conformation. Fig. 2 shows the projection of the crystal structure of (I) along the *b* axis. The molecules form layers with a thickness of c/2. In the layer, the long axes of the molecules are inclined to the *ab* plane. This layer structure is similar to that of the triclinic structure of the even-numbered *n*-alkanes containing 6–24 C atoms. It is considered that the arrangement in the layer is influenced by the steric and electrostatic repulsion of Br atoms at both ends. As a result, a molecular position in the layer slides along the direction of the long axis of the neighboring molecule.

Moreover, the layers are arranged in a zigzag manner between the neighboring layers, forming a herringbone motif similar to the tilt-smectic C phase and the smectic C_A phase of liquid crystals, as shown in Fig. 3. The layers are stacked closely in such a way that the α -CH₂ groups are allowed to fit into the grooves formed by Br atoms, with nearest contacts of 3.752 (3) Å, agreeing closely with the van der Waals contacts of 3.75 Å (Rowland & Taylor, 1996). Such a close packing is observed in the even-numbered alkane- α , ω -diols containing 4–19 and 21–23 C atoms (*e.g.* Thalladi *et al.*, 2000) and the α , ω dichloroalkanes containing 16, 20 and 26 C atoms (Takamizawa *et al.*, 1992).

organic papers

The features in the molecular and crystal structure of (I) are similar to those of the homologous series with an even number of C atoms, *viz.* 1,12-dibromododecane (Kulpe *et al.*, 1981), 1,14-dibromotetradecane (Uno & Nakamura, 2003), 1,16-dibromohexadecane (Kobayashi *et al.*, 1995), 1,18-dibromooctadecane (Nakamura *et al.*, 1993), and 1,26dibromohexadecane (Takamizawa *et al.*, 1992).

Experimental

Thin plate-like crystals of (I) were grown by slow evaporation of a solution containing a mixture of n-hexane and ethanol (1:1). The well developed face of the crystal is (001).

Crystal data

 $C_{20}H_{40}Br_2$ $D_x = 1.325 \text{ Mg m}^{-3}$ $M_r = 440.34$ Cu Ka radiation Monoclinic, $P2_1/n$ Cell parameters from 24 a = 5.482 (4) Å reflections b = 5.383 (2) Å $\theta = 9.5 - 15.5^{\circ}$ $\mu = 4.61 \text{ mm}^{-1}$ c = 37.412(2) Å $\beta = 91.50 \ (2)^{\circ}$ T = 296 (1) KV = 1103.6 (9) Å³ Plate, colorless Z = 2 $0.59 \times 0.47 \times 0.04 \text{ mm}$ Data collection Rigaku AFC-5R diffractometer $R_{\rm int} = 0.028$ $\theta_{\rm max} = 70.1^{\circ}$ ω –2 θ scans $h = -1 \rightarrow 6$ Absorption correction: by integration (NUMABS; $k = 0 \rightarrow 6$

 $l = -45 \rightarrow 45$

0.52]

 $(\Delta/\sigma)_{\rm max} < 0.001$ $\Delta \rho_{\rm max} = 0.38 \text{ e} \text{ Å}^{-3}$

 $\Delta \rho_{\rm min} = -0.62 \text{ e} \text{ \AA}^{-3}$

Extinction correction:

(Larson, 1970)

3 standard reflections

every 150 reflections

intensity decay: 6.8%

 $w = (4F_o^2)/[0.001F_o^2 + 5.3\sigma^2(F_o) +$

Extinction coefficient: 21.3 (3)

Higashi, 1999) $T_{min} = 0.325$, $T_{max} = 0.997$ 3004 measured reflections 2008 independent reflections 1667 reflections with $F^2 > 2\sigma(F^2)$

Refinement

Refinement on F^2 $R[F^2 > 2\sigma(F^2)] = 0.038$ $wR(F^2) = 0.116$ S = 1.002008 reflections 121 parameters H-atom parameters constrained

Table 1

Selected geometric parameters (Å, °).

Br1-C1	1.946 (4)	C6-C7	1.510 (5)
C1-C2	1.489 (5)	C7-C8	1.514 (4)
C2-C3	1.514 (5)	C8-C9	1.509 (5)
C3-C4	1.526 (5)	C9-C10	1.518 (4)
C4-C5	1.514 (5)	C10-C10 ⁱⁱ	1.507 (5)
C5-C6	1.522 (4)		
Br1-C1-C2-C3	-179.6(2)	C5-C6-C7-C8	179.2 (3)
C1-C2-C3-C4	179.4 (3)	C6-C7-C8-C9	179.8 (3)
C2-C3-C4-C5	179.7 (3)	C7-C8-C9-C10	-179.8(3)
C3-C4-C5-C6	-179.7(3)	C8-C9-C10-C10 ⁱⁱ	179.7 (3)
C4-C5-C6-C7	178.8 (3)	C9-C10-C10 ⁱⁱ -C9 ⁱⁱ	180.0 (3)

Symmetry code: (ii) 1 - x, -y, 2 - z.

Table 2

Intermolecular contacts (Å).

Br1···Br1 ⁱⁱⁱ	3.752 (3)	$Br1 \cdots Br1^{iv}$	3.752 (3)
Symmetry codes: (iii) $\frac{9}{2} - x$,	$y - \frac{1}{2}, \frac{3}{2} - z$; (iv)	$\frac{9}{2} - x, \frac{1}{2} + y, \frac{3}{2} - z.$	

H atoms were positioned geometrically and treated as riding, with C-H distances of 0.95 Å and $U_{iso}(H) = 1.2U_{eq}(parent)$.

Data collection: *MSC/AFC Diffractometer Control Software* (Molecular Structure Coorporation, 1992); cell refinement: *MSC/AFC Diffractometer Control Software*; data reduction: *Crystal-Structure* (Molecular Structure Coorporation & Rigaku, 2001); program(s) used to solve structure: *SIR*92 (Altomare *et al.*, 1994); program(s) used to refine structure: *CRYSTALS* (Watkin *et al.*, 1996); molecular graphics: *ORTEP-3 for Windows* (Farrugia, 1997); software used to prepare material for publication: *CrystalStructure*.

The authors express their gratitude to Mr K. Uno and Mr H. Shimizu for their support.

References

- Altomare, A., Cascarano, G., Giacovazzo, C., Guagliardi, A., Burla, M. C., Polidori, G. & Camalli, M. (1994). J. Appl. Cryst. 27, 435.
- Farrugia, L. J. (1997). J. Appl. Cryst. 30, 565.
- Higashi, T. (1999). NUMABS. Rigaku Corporation, Tokyo, Japan.
- Kobayashi, H., Yamamoto, T. & Nakamura, N. (1995). Cryst. Res. Technol. 30, 275–280.
- Kulpe, S., Seidel, I., Szulzewsky, K., Steger, U. & Steger, E. (1981). Cryst. Res. Technol. 16, 349–356.
- Larson, A. C. (1970). Crystallographic Computing, edited by F. R. Ahmed, S. R. Hall & C. P. Huber, pp. 291–294. Copenhagen: Munksgaard.
- Michaud, F., Ventolà, L., Calvet, M. T., Cuevas-Diarte, M. A., Solans, X. & Font-Bardía, M. (2000). Acta Cryst. C56, 219–221.
- Molecular Structure Corporation (1992). MSC/AFC Diffractometer Control Software. MSC, 3200 Research Forest Drive, The Woodlands, TX 77381, USA.
- Molecular Structure Corporation & Rigaku (2001). CrystalStructure. Version 3.10. MSC, 9009 New Trails Drive, The Woodlands, TX 77381-5209, USA, and Rigaku Corporation, 3-9-12 Akishima, Tokyo 196-8666, Japan.
- Nakamura, N., Uno, K. & Ogawa, Y. (2001). Acta Cryst. E57, o1091-01093.
- Nakamura, N., Yamamoto, T., Kobayashi, H. & Yoshimura, Y. (1993). Cryst. Res. Technol. 28, 953–957.
- Nyburg, S. C. & Gerson, A. R. (1992). Acta Cryst. B48, 103-106.
- Ogawa, Y. & Nakamura, N. (1999). Bull. Chem. Soc. Jpn, 72, 943–946.
- Rosen, L. S. & Hybl, A. (1972). Acta Cryst. B28, 610-617.
- Rowland, R. S. & Taylor, R. (1996). J. Phys. Chem. 100, 7384-7391.
- Takamizawa, K., Kodama, M., Matsunaga, S. & Shiokawa, K. (1992). Eng. Sci. Rep. Kyushu Univ. 13, 341–347.
- Thalladi, V. R., Boese, R. & Weiss, H. C. (2000). Angew. Chem. Int. Ed. 39, 918–922.
- Uno, K. & Nakamura, N. (2003). Acta Cryst. E59, o708-o710.
- Uno, K., Nakamura, N. & Ogawa, Y. (2002). Acta Cryst. A58 (Suppl.), C-338.
- Watkin, D. J., Prout, C. K., Carruthers, J. R. & Betteridge, P. W. (1996). *CRYSTALS*. Issue 10. Chemical Crystallography Laboratory, University of Oxford, England.